Course Code: 1910005 MLRITM- R19



## MARRI LAXMAN REDDY INSTITUTE OF TECHNOLOGY AND MANAGEMENT

## (AN AUTONOMOUS INSTITUTION)

(Approved by AICTE, New Delhi & Affiliated to JNTUH, Hyderabad)

Accredited by NBA and NAAC with 'A' Grade & Recognized Under Section2(f) & 12(B)of the UGC act,1956

## I B.Tech I Sem Supply End Examination, November 2020

CHEMISTRY (EEE, CSE, IT)

Time: 2 Hours. Max. Marks: 70

Note: 1. Answer any FIVE questions.

2. Each question carries 14 marks and may have a, b as sub questions.

1	a)	Discuss about LCAO theory. With the help of diagram explain Pi molecular orbitals of butadiene.	7M
	b)	Give the salient features of crystal field theory. What is the effect of doping on conductance? Explain.	7M
2		Discuss about crystal field splitting of transition metal "d "orbitals in tetrahedral and square planar geometries.	14M
3	a)	How do you estimate hardness of water by EDTA complexometry.	7M
	b)	Give the specifications for potable water. Write the steps involved in potable water.	7M
4		Differentiate primary and secondary batteries. Describe the construction, working and uses of Lithium ion battery.	14M
5	a)	Why ion-exchange process is superior over other methods of softening of water?	7M
	b)	What is electrochemical series? Write its applications.	7M
6		What is nucleophilic substitution reaction? Write the mechanism and stereochemistry of S N1 and SN2 reactions.	14M
7	a)	How do you synthesise aspirin. Give the pharmaceutical applications of aspirin and paracetamol.	7M
	b)	Define chemical shift. What is its significance? Write about MRI.	7M
8		Why do you study spectroscopy? Illustrate selection rules and applications of UV spectroscopy in quantitative analysis.	14M